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Synthesis of 7-Glycopyranosyl-5-oxq-pyrrolo[2,3-d]Pyrimidine and 4-Glyco Pyranosylamino-Furo[2,3-d]Pyrimidine Derivatives

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SYNTHESIS OF 7-GLYCOPYRANOSYL-5-OXO-PYRROLO[2,3-d]PYRIMIDINE AND 4-GLYCOPYRANOSYLAMINO-FURO[2,3-d]PYRIMIDINE DERIVATIVES

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<u>Abstract</u>: The synthesis of some glycosyl derivatives of furo[2,3-d] and pyrrolo[2,3-d] pyrimidines is described.

The reactions of 4-glycosylaminopyrimidine derivatives $\frac{1}{2}$ with chloracetyl chloride drive to the compounds $\frac{2}{2}$, $\frac{3}{2}$ and $\frac{4}{2}$, being the yields different depending on the solvents used (anhydrous chloroforme or ethyl a cetate). The compounds $\frac{2}{2}$ are the main product of reaction in ethyl aceta te whereas the compounds $\frac{3}{2}$ are the minority in both solvents and the yield of the compounds $\frac{4}{2}$ becomes higher in chloroforme.

Furo[2,3-d]pyrimidine derivatives $\underline{6}$ and pyrrolo[2,3-d]pyrimidine $\underline{5}$ were obtained by reaction of the compounds $\underline{2}$ in anhydrous $K_2^{CO}_3$ DMF solutions².

It has been proved a higher reactivity of the C_6 -OH group (which is present due to the tautomeric equilibrium whem R^1 = H) than C_A -NH group.

So, the reaction products are always furo[2,3-d]pyrimidine derivatives $\underline{6}$ and $\underline{7}$ when R^1 = H, whereas the pyrrolo[2,3-d]pyrimidine derivatives $\underline{5}$ are obtained when R^1 = CH $_3$.

All the compounds obtained were purified using column chromatography on silica gel and then by EtOH recrystallization. The proposed structures are in accordance with the spectroscopic data.

Using ¹H-NMR spectroscopy the hydrogen bridge bonds drawn in the scheme have been detected and likewise the existence of the tautomeric equilibria showed in the scheme was proved.

SCHEME 1

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